Maharaja Ranjit Singh College of Professional Sciences, Indore					
	Department of Biosciences				
	Lesson Plan - B. Sc. I Year Pharmaceutical Chemistry (July 2020 - June 2021)				
	Subject:Pharmaceutical Organic Chemistry (Paper -I)				
		Teacher - Dr. Mukesh Gupta			
	<b>T</b> T •4				
Day/Lecture	Unit	Торіс			
1	Unit 1	Historical development of Pharmaceutical chemistry,			
2		Atomic and molecular orbital			
3		Covalent bond, inter molecular forces,			
4		hybrid orbital, Bond dissociation energy(homolysis and heterolysis)			
5		Polarity of bonds and molecules, structure and physical properties			
6		Resonance			
7		hyperconjuction,			
8		hydrogen bonding, introduction, types,			
9		effect of hydrogen bonding			
10		Inductive effect, field effect			
11		Acids and bases, arrhenius concept			
12		Bronsted-Lowry concept,			
13		strength of acids and bases, Lewis concept,			
14		Bond dissociation energy			
15		pH,pKa,pKb Values,			
16		buffers, buffers in pharmaceutical			
17		buffers in biological system,Buffered isotonic solution			
18	Unit 2	Physicochemical properties and molecular constitution,			
19	Cint 2	surface and interfacial tention			
20		refractive index,optical rotation			
20		dielectric constant			
22		dipole moment, density, Viscosity			
23		molar refraction and parachor,			
23		stereo isomerism,			
24		Optical isomerism-optical activity			
25		Optical isomerism-optical activity			
20		enantiomerism.diastereoisomerism.			
18		meso compounds			
29		elements of symmetry			
30		DL system of nomenclature of optical isomers			
31		Chiral and achiral molecules			
32		RS system of nomenclature of optical isomers,			
33		Reaction of chiral molecules			
34		sequence rule			
35		Racemic modification and resolution of racemic mixture			
36		Geometrical isomerism, nomenclature of geometrical isomers			
37		Methods of determination of configuration of geometrical isomers			
38	Unit 3	Types of organic reaction,			
39		mechanism of organic reaction			
40		electrophiles and nucleophiles			
41		Curved arrow notation, drawing electron movement with arrow			
42		Half headed and double headed arrow,			
43		Reaction intermediates, formation, structure, stability and rectivity of carbocation			
44		Reaction intermediates, formation, structure, stability and rectivity of carbocation			
45		Reaction intermediates, formation, structure, stability and rectivity of carboanion			
46		Reaction intermediates, formation, structure, stability and rectivity of carboanion			
47		Reaction intermediates, formation, structure, stability and rectivity of free radical			
48		Reaction intermediates, formation, structure, stability and rectivity of free radical			
49		Nucleophilic aliphatic substitution			
50		SN1 and SN2 reaction, mechanism			
51		kinetics, order of reactivity and stereochemistry of nucleophile, Elimination reaction			
52		Elimination reaction			
55	1				

53		E1 and E2 reaction, mechanism
54		kinetics, order of reactivity and stereochemistry of electrophile
55		kinetics,order of reactivity and stereochemistry of electrophile
60	Unit 4	Classification of drugs on the basis of biological sources
61		Classification of drugs on the basis of Geographical sources
62		Classification of druge on the basis of Marine and Minerals sources
63		Theories of drug action, biological defenses
64		Theories of drug action, chemical defenses
65		Surface active agents, metabolic antagonism
66		Enzyme neutralizers, drug receptor interactions and receptor theories
67		drug receptor interactions and receptor theories
68	Unit 5	Introduction to dosage forms, classification and definition
69		Rout of drug administration
70		Aromatic waters and syrups
71		Tinctures and infusion
72		Introduction to medicinal system, Ayurvedic, Unani,
73		Introduction to medicinal systemSiddha
74		Introduction to medicinal system, Homeopathic, Allopthic
75		Weight and measures, Imperial and metric system
76		Calculation involving percentage solutions, allegation
77		Proof sprit and isotonic solution based on freezing point and molecular weight

Maharaja Ranjit Singh College of Professional Sciences, Indore					
Department of Biosciences					
Lesson Plan - B. Sc. I Year Pharmaceutical Chemistry (July 2020 - June 2021)					
	Subject:Inorganic Pharmaceutical analysis (Paper -II)				
Teacher - Dr. Mukesh Gupta					
Day/Lecture	Unit	Topic			
1	Unit 1	Impurities in pharmaceutical substances.			
2		history of pharmacopoeia			
3		Sources and types of impurities			
4		effect of impurities			
5		Permissible impurities in pharmaceutical substances			
6		Methods used to purify inorganic substances			
7		Test of purity, introduction of limit test			
8		Principle of limit test,limit test for chloride			
9		Limit test for Sulphate,			
10		Limit test for Iron			
10		Limit test for Arsenic,			
12		Limit test for Lead			
13		Limit test for Heavy metals			
13	Unit 2	Pharmaceutical analysis,			
15		different techniques of analysis			
16		Methods of expressing concentration			
10		Primary and secondary standard solution,			
18		preparation of solution			
10		Prepration and standardization of various molar and normal solution			
20		Oxalic acid,Sodium hydroxide,			
20		hydrochloric acid,Sodium hydroxide,			
21		SodiumsulphateSuphuric acid			
22		potassium permanganate and ceric ammonium sulphate			
23		Errors, sources of errors, types of errors			
24		Methods of minimizing errors			
25		Accuracy, precision and significant figures			
20	Unit 3	Acid base titration,			
	Unit 5	theories of acid base titrtion			
28					
29		Classification of acid base titration and theory involved			
30		titritionin strong acid and strongbase,			
31		titration weak acid and base			
32		titration weak acid and base			
33		Neutralization curves			
34		Non aqueous titration, solvents,			
35		acidimetry and alkalimetry tration			
36		Estimation of sodium benzoate and Ephedrine HCl			
37		Redox titration,			
38		concept of oxdidation and reduction			
39		Types of redox titration			
40		Principle and application of Cerimetry, Iodimetry			
41		Iodometry titration with potassium iodate			
42	Unit 4				
43		Mohr's method			
44		Volhrd's method			
45		sodium chlorideFajans method, estimation of			
46		complexometric titration, classification			
47		Metal ion indicators,			
48		masking and demasking reagents			

49		Estimation of Magnesium sulphate
50		Estimation of Calcium gluconate
51		Gravimetryanalysis
52		Principle, step involved in gravimetric analysis
53		Purity of precipatate,
54		co-precipitation and post precipitation
55		Estimation of barium sulphate
56		Basic principle, method and application of diazotisation titration
57	Unit 5	Preparation and uses of Alum
58		Preparation and uses of Aluminium hydroxide gel
59		Preparation and uses of Antimony potassium tartrate
60		Preparation and uses of Aromatic spirit of ammonia
61		Preparation and uses of boric acid
62		Preparation and uses of Potassium citrate
63		Praparation and uses of Sodium benzoate,
64		Praparation and uses of milk of Magnesia
65		preparation and uses of Magnesium carbonate,
66		preparation and uses of Zinc Oxide
67		
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Department of Biosciences

Lesson Plan - B. Sc. I Year Pharmaceutical Chemistry (July 2020 - June 2021) Subject - Pharmaceutical Chemistry Practical

		Teacher - Dr. Mukesh Gupta
Day/Lecture	Unit	Торіс
1		Identification of elements and groups present in organic compounds
2		Identification of elements and groups present in organic compounds
3		Identification of elements and groups present in organic compounds
4		Identification of elements and groups present in organic compounds
5		Identification of elements and groups present in organic compounds
6		Determination of solubility of benzoic aqcid over a range of temperature
7		Determination of surface tention of the given liquids
8		Determination of Viscosity of the given liquids
9		Preparation of aromatic Waters
10		preparation of Syrup
11		Preparation of Tinctures
12		Preparation of buffer solutions and measurement of pH
13		Identification of the unknow compoundfrom the literature using MP/BP.
14		Limit test of chloride
15		Limit test of Sulphate
16		Limit test of Iron
17		Limit test of Lead
18		Preparation of inorganic pharmaceutical Alum
19		Preparation of inorganic pharmaceutical Aluminium hydroxide gel
20		Preparation of inorganic pharmaceutical milk of magnesia
21		Preparation of inorganic pharmaceutical ferrous ammonium sulphate
22		Preparation of inorganic pharmaceutical antimony potassium tartarte
23		Preparation and standardization of sodium hydroxide, Oxalic acid
24		Assay of Ammonium chloride
25		Assay of borax
26		Assay of Zinc Oxide
27		Assay of Sodium carbonate
28		Assay of Copper Sulphate by Iodometry
29		Volumetric estimation of ferrous sulphate using oxalic acid,
30		Potassium permanagnate and potassium dichromate.
31		
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33 34		
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	Mahara	aja Ranjit Singh College of Professional Sciences, Indore			
		Department of Chemical Science			
	Lesson Pl	an - B.Sc. II Year Pharmaceutical Chemistry(July 2020 - June 2021)			
	Subject - Medicinal chemistry Paper I				
		Teacher - Dr Mukesh Gupta			
ay/Lectu	Unit	Topic			
1	Unit 1	Physicochemical properties in relation tobiological action			
2	Chit I	Ionization, solubility, partition coefficint			
3		Hydron bonding-introduction,types,condition for hydron bonding			
4		effect of hydron bonding ,examples			
5		protien binding introduction, definition, examples,			
6		protien binding applications			
7		chelation introduction, definition, examples			
8		importance of chelation			
9		bioisosterism			
10		optical and geometrical isomerism introduction, classification			
11		optical isomerism introduction, reason of optical isomerism, examples			
12		geometrical isomerism introduction, reason of geometrical isomerism			
13		geometrical isomerism in various compounds			
14	Unit 2	General Anesthetics: Definition, Stages of Anesthesia			
15		Classification and Theories of General Anesthetics			
16		Mechanism of action of general anesthetics			
17		Synthesis of nitrous oxide			
18		Synthesis of halothane			
19		Synthesis of thiopental sodium			
20		Synthesis of Chloroform			
21		local anesthetics introduction			
22		local anesthetics classification			
23		mechainsm of action of local anesthetics			
24		SAR of local anesthetics			
25		synthesis of procaine hydrochloride			
26		synthesis of procaine hydrochloride			
27		synthesis of Benzocain			
28		synthesis of Benzocain			
29		synthesis of lignocaine hydrochloride			
30		synthesis of lignocaine hydrochloride			
31	Unit 3	Hypnotics and Sedatives introduction, examples			
32		Definition and Classification of hypnotics and sedatives			
33		Mechanism of action of hypnotics and sedatives			
34		SAR of Barbituric acid derivatives			
35		Sybthesis of Barbital			
36		Sybthesis of Barbital			
37		Sybthesis of allobarbital			
38		Sybthesis of allobarbital			
39		Sybthesis of hexobarbital			
40		Sybthesis of hexobarbital			
41		SAR of benzodiazepines			

42		Synthesis of diazepam
43		Synthesis of diazepam
44		Synthesis of alprazolam
45		Synthesis of alprazolam
46		Synthesis of zolpidem
47		Synthesis of zolpidem
48		Anti-convulsants introduction, definition, examples etc.
49		classification of anti-convulsants
50		mechanism of action of anti-convulsants
51		Synthesis of phenobarbital
52		Synthesis of phenobarbital
53		Synthesis of phenytoin sodium
54		Synthesis of phenytoin sodium
55	Unit 4	Analgesics and antipyretics introduction, classification
56		Mechanism of action and SAR of morphine analogue
57		Mechanism of action and SAR of Salicylic acid
58		Mechanism of action and SAR of aryl alkanoic acid derivatives
59		Synthesis of Aspirin
60		Synthesis of paracetamol
61		Anti-histaminics drugs introduction, classification
62		Mechanism of action and SAR of ethanolamine derivatives
63		Synthesis of diphenhydramine hydrochloride
64		Synthesis of diphenhydramine hydrochloride
65		Synthesis of promethazine hydrochloride
66		Synthesis of promethazine hydrochloride
67	Unit 5	Diuretics introduction, classification
68		Mechanism of action and SAR of diuretics
69		Synthesis and uses of hydrochlorthiazide
70		Synthesis and uses of hydroflumethiazide
71		Synthesis and uses of ethacrynic acid
72		Synthesis and uses of furosemide
73		Synthesis and uses of acetazolimide
74		Antihypertensives drugs introduction, classification
75		Mechanism of action and SAR of antihypertensive drugs
76		Synthesis of captropril
77		Synthesis of propranolol hydrochloride

Department of Chemical Science

Lesson Plan - B.Sc. II Year Pharmaceutical Chemistry(July 2020 - June 2021)

Subject - Chemistry of Natural Products Paper II

## Teacher - Dr Mukesh Gupta

Day/Lecture	Unit	Topic
1	Unit 1	Heterocyclic Compounds
2		Nomenclature,
3		Structural formula and chemistry of Imidazoles,
4		Structural formula and chemistry of Imidazoles
5		Structural formula and chemistry of Oxazoles
6		Structural formula and chemistry of Oxazoles
7		Structural formula and chemistry of Pyrazoles
8		Structural formula and chemistry of Pyrazoles
9		Structural formula and chemistry of Pyran
10		Structural formula and chemistry of Pyran
11		Structural formula and chemistry of Pyrimidine,
12		Structural formula and chemistry of Pyrimidine,
13		Structural formula and chemistry of Indole
14		Structural formula and chemistry of Indole,
15		Structural formula and chemistry of Isoquinoline
16		Structural formula and chemistry of purine
17		Structural formula and chemistry of pirine
18	Unit2	Carbohydrates: Classification of Carbohydrates
19		Monosaccharides
20		Glucose, Fructose and their reactions
21		Glucose, Fructose and their reactions
22		Cyclic structure of D-glucose
23		Mutarotation. Diasaccharides
24		Maltose
25		Lactose,
26		Sucrose.
27		Polysaccharides : Starch,
28		Polysaccharides : Starch
29		Polysaccharides : Cellulose
30		Polysaccharides : Cellulose
31		Polysaccharides : dextran,
32		Polysaccharides : dextran
33		Polysaccharides : glycogen
34		Polysaccharides : glycogen
35		Polysaccharides : insulin
36		Polysaccharides : insulin
37		Fats,Oils,Waxes,fatty acids
38		Physico-chemical properties
39		Phospholipids
40		lecithenes
41		cephalins
42		plasmogenes
43		glycolipids

44	Unit 3	Amino acids, classification
45		Structure and stereochemistry of aminoacids
46		properties of amino acids
47		properties of amino acids
48		Protein, Classification
49		properties of proteins
50		primary secondary and tertiary structure of proteins
51		primary secondary and tertiary structure of proteins
52		Nucleic acids introduction
53		Structure of DNA and RNA
54	Unit 4	Alkaloids : Classification
55		general introduction, composition
56		chemistry and chemical classes, biosources
57		Therapeutic uses and commercial application of Quinine
58		Therapeutic uses and commercial application of Quinine
59		Therapeutic uses and commercial application of morphine
60		Therapeutic uses and commercial application of morphine
61		Therapeutic uses and commercial application of reserpine
62		Therapeutic uses and commercial application of reserpine
63		Glycosides: Classification
64		general introduction, composition
65		chemistry and chemical classes, biosources
66		Therapeutic uses and commercial application of senna
67		Therapeutic uses and commercial application of senna
68		Therapeutic uses and commercial application of aloes
69		Therapeutic uses and commercial application of aloes
70		Therapeutic uses and commercial application of bitter almond
71		Therapeutic uses and commercial application of bitter almond
72	Unit 5	Terpenes:Classification. Isolation
73		general introduction, composition
74		chemistry and chemical classes, biosources
75		Therapeutic uses and commercial application of citral
76		Therapeutic uses and commercial application of citral
77		Therapeutic uses and commercial application of carvone
78		Therapeutic uses and commercial application of carvone
79		Therapeutic uses and commercial application of menthol
80		Therapeutic uses and commercial application of m enthol
81		Therapeutic uses and commercial application of thymol
82		Therapeutic uses and commercial application of
83		Steroides : Isolation, Nomenclature
84		Chemistry of Cholesterol
85		Chemistry of Cholesterol
86		Chemistry of ergosterol
87		Chemistry of ergosterol
88		Chemistry of stigmasterol
89		Chemistry of stigmasterol
90		Chemistry of cartosone
91		Chemistry of cartosone
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Department of Chemical Science

Lesson Plan - B.Sc. II Year Pharmaceutical Chemistry(July 2020 - June 2021)

Subject - Pharmaceutical Practical

## Teacher - Dr Mukesh Gupta

Day/Lecture	Unit	Торіс
1		Purification of pharmaceutical organic compounds:Decolorization,recrystallization,sublimation
2		Purification of pharmaceutical organic compounds:Decolorization,recrystallization,sublimation
3		Purification of pharmaceutical organic compounds:Decolorization,recrystallization,sublimation
4		Preparation of Benzocaine
5		Preparation of phenttoin
6		Preparation of aspirin
7		Preparation of paracetamol
8		Determination of partition coefficient for any two drugs
9		Isolation of caffeine from tea
10		Isolation of casein from milk
11		Determination of Iodine value
12		Determination of acid value
13		Determination of saponification value
14		Separation of amino acids by npaper chromatography
15		Identification test of carbohydrate, proteins
16		Separation of sugars by thin layer chromatography
17		Separation of plant pigments by column chromatography
18		Synt5hesis of benzyl
19		Synt5hesis of thalimide
20		Synt5hesis of sulphanic acid

	Mahara	aja Ranjit Singh College of Professional Sciences, Indore				
		Department of Chemical Sciences				
	Lesson Plan - B.Sc. III Year Pharmaceutical Chemistry (July 2020 - June 2021)					
		Subject - Medicinal Chemistry Paper I				
		Teacher - Dr. Mukesh Gupta				
ay/Lectu	Unit	Торіс				
1	Unit 1	Adrenergic- classification				
2		Mechanism of action and SAR of ethylamine analogue				
3		Mechanism of action and SAR of ethylamine analogue				
4		Synthesis of adrenalin				
5		Synthesis of epinephrine				
6		Synthesis of norepineprine				
7		Synthesis of ephedrine				
8		Synthesis of dopamine				
9		Anticoagulants-classification				
10		Mechanism of action				
11		Synthesis and uses of haparin				
12		Synthesis and uses of haparin				
13		Synthesis and uses of dicoumarol				
14		Synthesis and uses of dicoumarol				
15		Expectorants and anti-tussives, classification				
16		Mechanism of action				
17		Synthesis of acetylcystein				
18		Synthesis of guaifensin				
19		Synthesis of noscapine				
20	Unit 2	Antibiotics, historical background				
21		Structure activity relationship				
22		Chemical classification of beta-lactam antibiotics				
23		Penicillin				
24		cepholosporins				
25		Aminoglycosides				
26		Streptomycin and neomycin				
27		Tetracyclines: tetracycline				
28		macrolide				
29		Azithromycin				
30		Constitution and synthesis of chloramohenicol				
31		Constitution and synthesis of chloramohenicol				
32		Sulphonamide: classification				
33		Mechanism of action				
34		Synthesis ans uses of suphacetamide				
35		Synthesis ans uses of suphacetamide				
36		Synthesis ans uses of sulphaguianide				
37		Synthesis ans uses of sulphaguianide				
38		Synthesis ans uses of Dapsone				
39		Synthesis ans uses of Dapsone				
40	Unit 3	Anti-malarial: classification				
41		Mechanism of action				

42		SAR of 4-aminoquinolines
43		Synthesis of chloroquine phosphate
44		Synthesis of chloroquine phosphate
45		Synthesis of emologume phosphate Synthesis of amidiquine hydrochloride
46		Synthesis of amidiquine hydrochloride
47		Synthesis of primquine phosphate
48		Synthesis of primquine phosphate
49		Anti-tubercular drugs: classification
50		Mechanism of action
51		Synthesis and uses of para amino salicylic acid
52		Synthesis and uses of para amino salicylic acid
53		Synthesis and uses of Isoniazid
54		Synthesis and uses of Isoniazid
55		Synthesis and uses of rifampicin
56		Synthesis and uses of rifampicin
57		Anti-amoebic: classification
58		Mechanism of action
59		Synthesis and uses of metronidazole
60		Synthesis and uses of metronidazole
61	Unit 4	Anti-diabetic : classification
62		Mechanism of action
63		Synthesis and uses of metformin
64		Synthesis and uses of metformin
65 66		Synthesis and uses of sitagliptin
67		Synthesis and uses of sitagliptin
68		Synthesis and uses of glimperide   Synthesis and uses of glimperide
69		Anti-neoplastic: types of cancer, classification
70		Mechanism of action
70		Synthesis and uses of 5-fluoro uracil
72		Synthesis and uses of 5-fluoro uracil
73		Synthesis and uses of 6-mercaptopurine
74		Synthesis and uses of 6-mercaptopurine
75		Synthesis and uses of thiotepa
76		Synthesis and uses of thiotepa
77		Synthesis and uses of busulphan
78		Synthesis and uses of busulphan
79	Unit 5	Drug design : Various approaches used in drug design
80		Physicochemical parameters used in quantitative structure activity
81		Physicochemical parameters used in quantitative structure activity
82		Relationship (QSAR) such as partition coefficient
83		Hammet's electronic parameter
84		Tafts steric parameter and Hansch analysis
85		Tafts steric parameter and Hansch analysis
86		Free Wilson analysis
87		introduction to 3D QSAR approaches
88		introduction to 3D QSAR approaches

	Mahara	ja Ranjit Singh College of Professional Sciences, Indore
		Department of Chemical Sciences
	Lesson Pla	n - B.Sc. III Year Pharmaceutical Chemistry (July 2020 - June 2021)
		Subject - Instrumental Chemistry Paper II
		Teacher - Dr. Mukesh Gupta
Day/Lecture	Unit	Topic
1	Unit 1	Conductometry- introduction, conductivity cell
2	Unit I	Conductometric titrations, applications
3		Potentiometry- Electrochemical cell
4		Construction and working of reference Standard hydrogen electrode
5		
6		Construction and working of reference Silver chloride electrode Construction and working of reference Calomel electrode
-		
7		Indicator electrode- metal electrode
8		Indicator electrode- Glass electrode
9		Methods to determine end point of potentiometric titration and application
10		Methods to determine end point of potentiometric titration and application
11		Polarography- principle
12		Ilkovik equationj
13		Construction and working of dropping mercury electrode, application
14		Construction and working of rotating platinum electrode, application
15	Unit 2	UV Spectroscopy- Principle
16		Instrumentation of UV Spectroscopy
17		Application of UV Spectroscopy
18		IR Spectroscopy- Principle
19		Instrumentation of IR Spectroscopy
20		Application of IR Spectroscopy
21	Unit 3	NMR Spectroscopy- Principle
22		The spinning nucleus
23		Magnetic and non-magnetic nuclei
24		Rules to find nuclear spin
25		Effect of external magnetic field
26		Professional motion and frequency
27		Chemical shift: measurment
28		Shielding and deshielding effect
29		Factors affectinf spin-spin coupling
30		Coupling constant
31		Instrumentation of NMR spectroscopy
32		Interpretation of NMR spectra
33		
		Mass spectroscopy : principle, instrumentation
34		Types of ions and use of mass spectrometry in determination of molecular weight
35	TT:+ 4	Types of ions and use of mass spectrometry in determination of molecular weight
36	Unit 4	Chromatography: introduction to chromatography
37		Adsorption and partition column chromatography- methodology
38		Adsorption and partition column chromatography-advantage, disadvantage
39		Adsorption and partition column chromatography-application
40		Thin layer chromatography-introduction, principle
41		Thin layer chromatography-methodology,Rf-values
42		Thin layer chromatography-advantage, disadvantage and application
43		Paper chromatographyintroduction, principle
44		Paper chromatography-methodology, development techniques
45		Paper chromatography-advantage, disadvantage and application
46	Unit 5	Gas chromatography-introduction, theory
47		Instrumentation of Gas chromatography
48		Derivatization temperature programming
49		Gas chromatography- advantage, disadvantage and application
50		High performance liquid chromatography (HPLC)-introduction, theory
51		High performance liquid chromatography (HPLC)- Instrumentation
52		High performance liquid chromatography (HPLC)- advantage and application

Department of Chemical Sciences

Lesson Plan - B.Sc. III Year Pharmaceutical Chemistry (July 2020 - June 2021)

Subject - Pharmaceutical Chemistry Practical

Teacher - Dr. Mukesh Gupta			
Day/Lecture	Unit	Торіс	
1	Unit 1	Preparation of N-phenyl azo beta- naphthol	
2		Preparation of diphenyl thiourea	
3		Preparation of methyl orange	
4		Preparation of phenolphthalein	
5		Interpretation of given UV spectrum	
6		Interpretation of given IR spectrum	
7		Determination of absorption maxima and effect of solvents on absorption maxima of Org.copm.	
8		Assay of Paracetamol by UV spectroscopy	
9		Estimation of dextrose by colorimetry	
10		Weight variation test of given tablets	
11		Hardness test of given tablets	
12		Friability test of given tablets	
13		Disintegration test of given tablets	
14		Systematic separation and identification of organic binary mixture	
15		Systematic separation and identification of organic binary mixture	
16		Systematic separation and identification of organic binary mixture	
17		Systematic separation and identification of organic binary mixture	
18		Systematic separation and identification of organic binary mixture	
19		Separation of mixture of amino acid and determination of Rf value by thin layer chromatography	
20		Separation of mixture of amino acid and determination of Rf value by thin layer chromatography	
21		Demonstration experiment on HPLC	
22		Demonstration experiment on Gas chromatography	